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red phosphorus red phosphorus is an allotrope of phosphorus it is an amorphous polymeric red solid that is stable in air it can be easily converted from white phosphorus under light or heating

red phosphorus may be formed by heating white phosphorus to 300 c 570 f in the absence of air or by exposing white phosphorus to sunlight red phosphorus exists as an amorphous network

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dec 14 2018   to obtain red p you must heat the white p obtained in this process and hold the temperature constant for an extended period the degradation from white to red phosphorus can

with the help of kindhearted donors amazing tech savvy volunteers and corporate support from microsoft and techno brain red p is in operation in usa and kenya

red phosphorus is one of the most common allotropes of phosphorus and is considered to be a derivative of the p 4 molecule it exists in an

amorphous non crystalline network of phosphorus atoms

what is red phosphorus in chemistry a red phosphorus is a stable non toxic allotrope of the element phosphorus it is commonly used for making safety matches fireworks and as a chemical reagent in

red phosphorus is the most common allotrope of phosphorus due to its relative stability compared to white phosphorus it has a variety of uses including but not limited to use as a catalyst battery

red phosphorus is produced by heating white phosphorus to high temperatures in the absence of oxygen causing it to undergo an allotropic transformation unlike white phosphorus red phosphorus

oct 16 2022   red phosphorus also called red p is an amorphous red solid at standard conditions and is considerably more stable than the white form it is formed by heating white phosphorus to 250

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